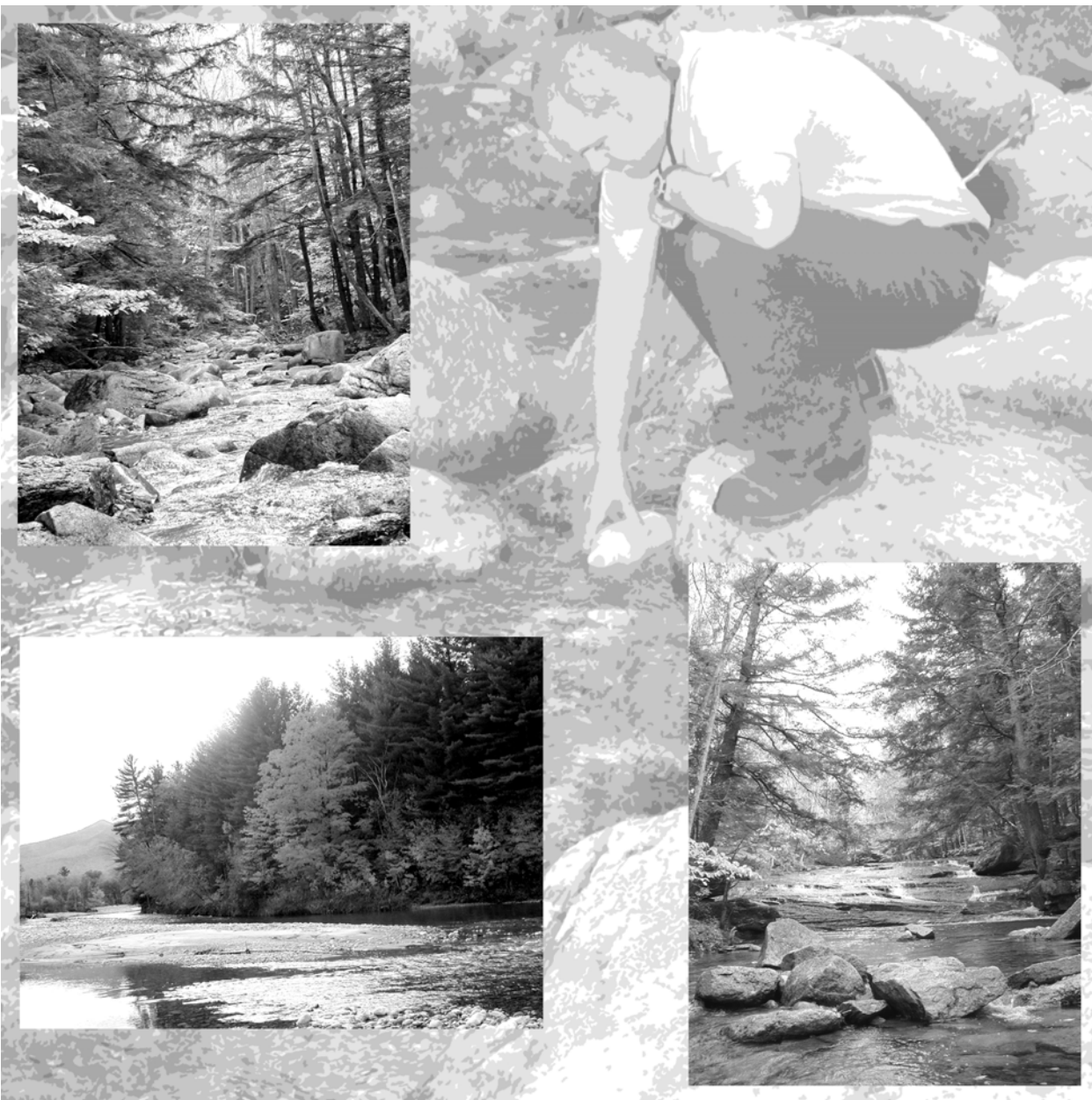




Database for Chemical Contents of Streams on the White Mountain National Forest

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Abstract

Producing and protecting high-quality streamwater requires background or baseline data from which one can evaluate the impacts of natural and human disturbances. A database was created for chemical analyses of streamwater samples collected during the past several decades from 446 locations on the White Mountain National Forest (304,000 ha in New Hampshire and Maine). The database is described and examples illustrating how the data can be used to examine ranges, trends, and relationships are included.

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Introduction

Producing and protecting high-quality water has always been a management goal on the White Mountain National Forest (WMNF). Stimulus for this goal began with the Organic Act of 1897 and the Weeks Law of 1911, which led to purchase of land for National Forests to “secure favorable conditions of water flows,” and “protect the watersheds of navigable streams.” This mandate was strengthened by requirements of the Resources Planning Act, the National Forest Management Act, the National Environmental Protection Act, and the Clean Water Act.

Producing high-quality water requires background or baseline data from which one can evaluate the impacts of natural and human disturbances. We created a database for chemical analyses of stream samples collected over the past several decades from managed watersheds on the WMNF that were not subjected to recent disturbance such as harvesting or land development. In this report we describe the database and present some examples that illustrate how the data can be used to examine ranges, trends, and relationships. The database is available free of charge from the WMNF.

White Mountain National Forest

The WMNF (304,000 ha) is located in central New England in the heart of the Appalachian-New England Mixed Forest-Coniferous Forest-Alpine Meadow Ecological Province (McNab and Avers 1994). The WMNF is well watered with an average annual precipitation of 1,100 mm (about one-third is snowfall) that is distributed uniformly throughout the year. About 60 percent of annual precipitation becomes streamflow, giving rise to several thousand kilometers of perennial streams that are highly valued for recreation, municipal, and commercial purposes. The chemical quality of these streams is affected by factors such as vegetation, geologic and soil characteristics, biological processes, and natural and human disturbances.

Vegetation

About two-thirds of the WMNF supports forests capable of timber production. Northern hardwoods, dominated by some combination of sugar maple, yellow birch, and American beech, is the dominant forest type. Northern conifers, including mixtures of balsam fir, red spruce and hemlock, are dominant in rocky and sandy habitats and at elevations above 775 m. At elevations above 1400 m (approximately tree line), vegetation grades from montane spruce-fir to alpine krummholz and alpine meadow.

Below tree line, most watersheds on the WMNF are forested with mixed types, making it difficult to determine whether one forest type influences stream chemistry more than another. However, all forest types on the WMNF, if not recently disturbed, have tight nutrient cycles, and leaching of nutrient ions to streams is minimal (Hornbeck et al. 1997). Recent findings suggest that stand aging may reduce nutrient concentrations of streamwater (Martin et al 2000a).

Geologic and Soil Characteristics

Bedrock underlying the WMNF includes Cambrian to Devonian sedimentary rocks that have been metamorphosed to quartzites, phyllites, schists, and gneiss, as well as a variety of intrusive and extrusive igneous rocks of Ordovician, Devonian, and Mesozoic age. Bedrock is mostly mantled by Wisconsin-age till deposited about 13,000 years ago during the last retreat of continental glaciation. Glaciofluvial and glaciolacustrine sediments are present in valleys. Otherwise, the till is unconsolidated and varies widely in composition and depth.

Soils developed in these glacial parent materials are primarily Haplorthods, Haplaquepts, and Dystrochrepts with sandy loam textures and high infiltration capacities. Mineralogical content of soils is highly variable due to the variety of lithologies present in

bedrock and transported parent materials (Bailey and Hornbeck 1992). The variability in mineralogical content and accompanying weathering rates is partly responsible for the range in alkalinity, acidity, and concentrations of base cations and aluminum found in streamwater within the WMNF.

Soil depths range from zero on exposed ledge to less than 0.1 m on ridge tops, to more than 1.0 m on benches, lower slopes, and valley bottoms. On upper slopes and at higher elevations, hydrologic flow paths and residence times for soil solution generally are short, lessening opportunities for chemical change before precipitation becomes streamflow. Deeper soils on lower slopes, benches, and valley bottoms provide for longer residence times for soil solution and greater opportunities for chemical and biological processes to both add to and subtract from the chemical content of the soil solution that eventually reaches streams (Johnson et al. 2000).

Biological Processes

Because streams on the WMNF tend to be cool, chemically dilute, shallow, and fast moving, biological processes and activities within streams are at low levels and have relatively minor impact on stream chemistry. Biologically mediated processes within terrestrial ecosystems (e.g. mineralization and nitrification) can affect chemical content of soil solution. However, as mentioned earlier, plant uptake and tight nutrient cycles on the WMNF limit leaching losses of nutrient ions generated by these processes, absent recent disturbances.

Natural and Human Disturbances

Natural disturbances on the WMNF include those from wind, ice, fire, insects and diseases, and beaver dams. Human disturbances include forest harvesting, recreational use and development, road and trail construction, and atmospheric deposition. If the disturbance is severe enough to adversely effect forest soils and/or disrupt plant nutrient uptake and cycling, there may be an increase in leaching losses of chemicals to streams. For example, intensive harvesting can elevate concentrations of base cations and nitrate in streams for 3 to 4 years after harvest (Hornbeck et al. 1987; Martin et al. 2000b). Atmospheric deposition can acidify streamwater and elevate concentrations of base cations, nitrate, sulfate, and aluminum. Such impacts range from severe at higher elevations with shallow soils to minimal on watersheds with abundant, base-rich minerals and buffered soils (Lawrence and Huntington 1999; Hornbeck et al. 1997). The database described in this report does not include samples collected from streams draining recent harvests (within the past 10 years) or those influenced by urban development.

The Database

Figure 1 shows locations of 446 sample stations from which stream chemistry data were obtained and entered into the database. Some of the stream sample stations are located outside the boundary of the WMNF. In such cases, a substantial portion of the upstream watershed is within the WMNF. The following shows the sources of stream data and the number of sample stations for each source.

<u>Source</u>	<u>Number of stations</u>
USDA Forest Service, Region 9	201
U.S. Geological Survey	25
U.S. Environmental Protection Agency, Region 1	48
U.S. Environmental Protection Agency, Headquarters	20
Maine Department of Environmental Protection	4
Institute of Ecosystem Studies	2
Northeastern Research Station	54
Christy Goodale (University of New Hampshire) and Peter Vitousek (Stanford University)	92
<u>Total</u>	<u>446</u>

Samples included in the database were collected between 1953 and the end of 1999. The number of samples collected at any given site ranged from fewer than 5 to more than 1,000; sampling intervals ranged from weekly to sporadic. The number of chemical determinations for a sample ranged from one to many depending on the objectives and analytical and financial resources available. Analyses were conducted by numerous laboratories and methods. The laboratories were of high quality and used current methods and equipment. In keeping with continuing improvements in analytical equipment and methods, more recent data may be more accurate and precise than earlier data.

The final database included more than 86,000 chemical determinations divided among 25 chemical parameters (Table 1). To facilitate comparisons, hydrogen ion was stored in the database as both pH and mg/L. Concentrations of all other elements and compounds were stored as ug/L or mg/L; conductivity was stored as microsiemens.

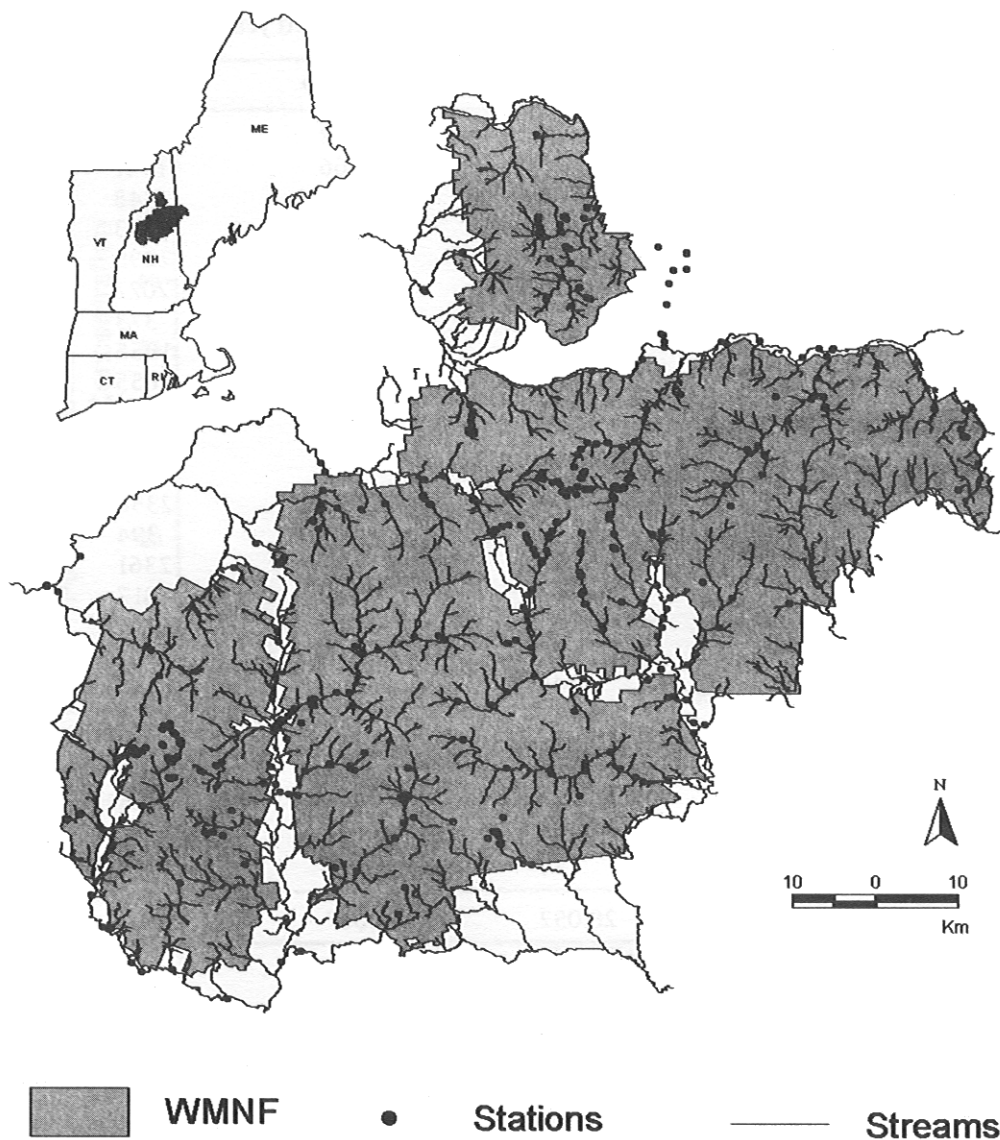


Figure 1.—Locations of sampling stations on the White Mountain National Forest.

Microsoft® Access 97¹, a relational database, was used for storing and analyzing the stream chemistry data. A substantial amount of the data for WMNF, especially that collected by federal agencies, was downloaded into Microsoft® Excel and then Access 97 from the STORET (short for STORage and RETrieval) database for water chemistry maintained by the U.S. Environmental Protection Agency (www.epa.gov/OWOW/STORET/index.html). Additional information for the database were obtained from literature searches, reports, and theses.

Data are stored as an Access file titled "WMNF Stream Chemistry." Associated files contain descriptions of fields and formats used to store data and individual station variables (latitude, longitude, elevation, etc.), and information on the laboratory methods used to obtain chemical data. The "WMNF Stream Chemistry" file is available upon request from:

Watershed Specialist, Forest Supervisor's Office
White Mountain National Forest
719 Main Street
Laconia, NH 03246-0772

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Use of the WMNF Stream Chemistry file requires a working knowledge of Microsoft® Access 97. The requested file will not contain data from the Hubbard Brook Experimental Forest located on the WMNF. Data for Hubbard Brook streams can be obtained from a web

Table 1.—Number of samples in database by chemical parameter and year

Parameter	Before 1970	1970-79	1980-89	1990-99	Total
Aluminum	212	374	1243	1526	3355
Ammonium	268	1138	966	1941	4313
Antimony	0	0	3	348	351
Arsenic	1	16	80	373	470
Calcium	368	3424	1449	2299	7540
Chloride	288	2704	1496	2077	6565
Chromium	0	82	99	373	554
Conductivity	74	2112	1695	1955	5836
Copper	7	149	342	655	1153
Iron	16	386	654	1265	2321
Lead	2	141	98	372	613
Manganese	24	64	339	720	1147
Magnesium	345	2576	1433	2358	6712
Nickel	2	23	67	394	486
Nitrate	272	3108	1454	2361	7195
pH	279	2795	1941	2817	7832
Potassium	388	2462	1045	2254	6149
Selenium	0	14	74	391	479
Silica	251	648	809	1280	2988
Silver	0	2	50	375	427
Sodium	388	2652	1437	2345	6822
Sulfate	262	1056	1380	2237	4935
Alkalinity	59	280	1232	1093	2664
Phosphate	74	1692	1345	1251	4362
Zinc	6	154	276	644	1080
Total	3586	28,052	21,007	33,704	86,349

site (www.hbrook.sr.unh.edu), which contains special instructions for using and citing data.

Using the Database

The stream-chemistry data can be used to examine a variety of subjects, including ranges, trends, and relationships, and to determine how chemical data from individual locations compare with WMNF-wide data. Rather than address the many possible applications for the database, we choose calcium (Ca^{2+}) concentrations of streamwater to illustrate potential applications because Ca^{2+} is the dominant cation in most streams, has been measured frequently, and is a concern with respect to nutrient depletion and forest health (Hornbeck et al. 1997; Lawrence and Huntington 1999).

Cumulative Frequency Plots

The use of cumulative frequency plots with mean concentration for each location (Figs. 2-3) is an effective way to display the range in values for the WMNF and to determine where individual locations fall within that range. The range in mean Ca^{2+} concentrations of 348 stations that were sampled for Ca^{2+} is ~0 to 240 $\mu\text{eq/L}$;

and the median concentration is slightly more than 75 $\mu\text{eq/L}$ (Fig. 2). About 95 percent of the 348 stations have mean Ca^{2+} concentrations less than 150 $\mu\text{eq/L}$. Closer examination of the database shows that the low end of the range (less than 50 $\mu\text{eq/L}$) usually include streams that drain small or high-elevation watersheds, while mean values at the higher end of the range include larger river basins. Cumulative frequency plots for pH and other ions including potassium, sodium, magnesium, aluminum, chloride, sulfate, and nitrate are shown in the Appendix.

Figure 3 shows the same cumulative frequency plot for Ca^{2+} but with one standard deviation about each mean. The plot shows that variation in Ca^{2+} concentrations at individual sampling locations is relatively small, particularly at the lower end of the range. The relatively small range and variability for Ca^{2+} in streamwater is indicative of low weathering rates and tight nutrient cycles across the WMNF.

Relationships to Elevation

A plot of mean Ca^{2+} concentrations versus elevation (Fig. 4) suggests that concentrations decrease with increasing elevation, though the relationship is not especially

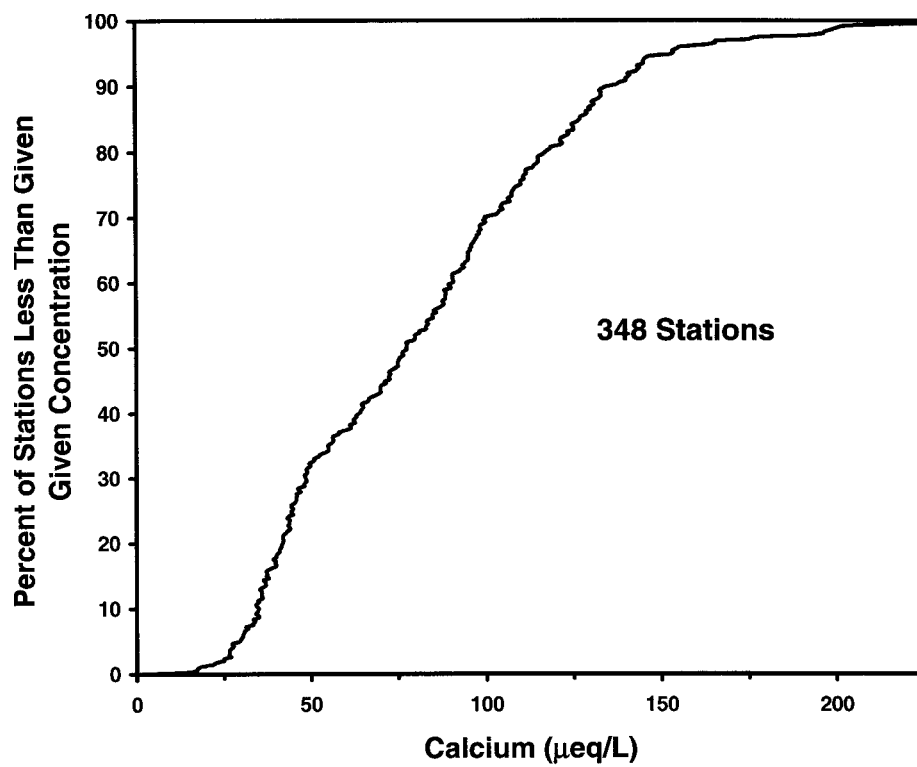


Figure 2.—Cumulative frequency plot for mean Ca^{2+} .

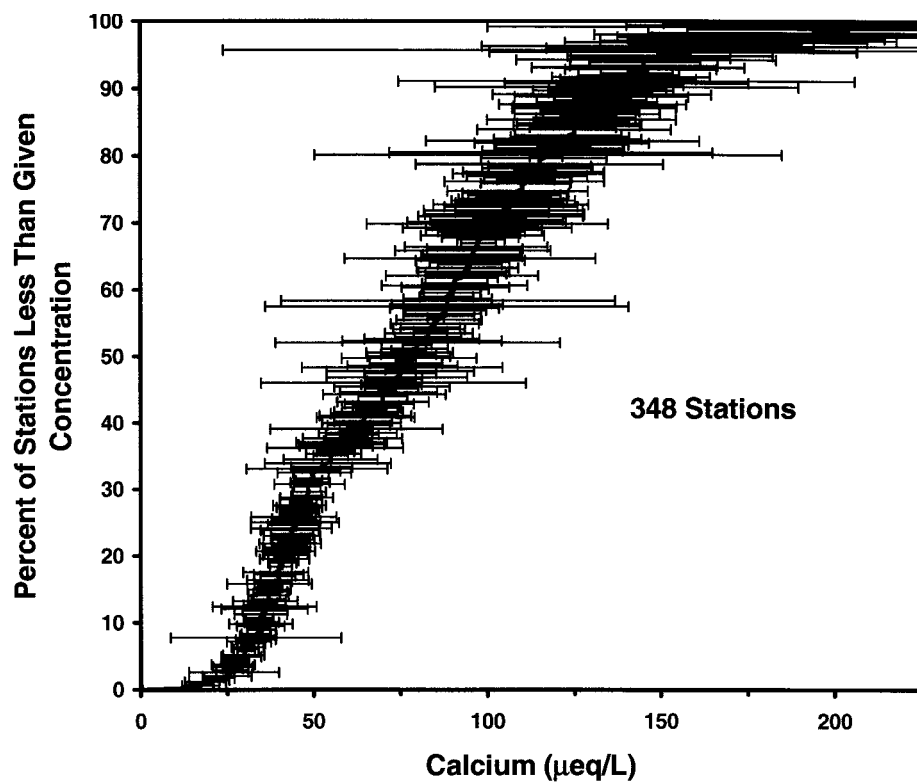


Figure 3.—Cumulative frequency plot for Ca^{2+} with one standard deviation about mean.

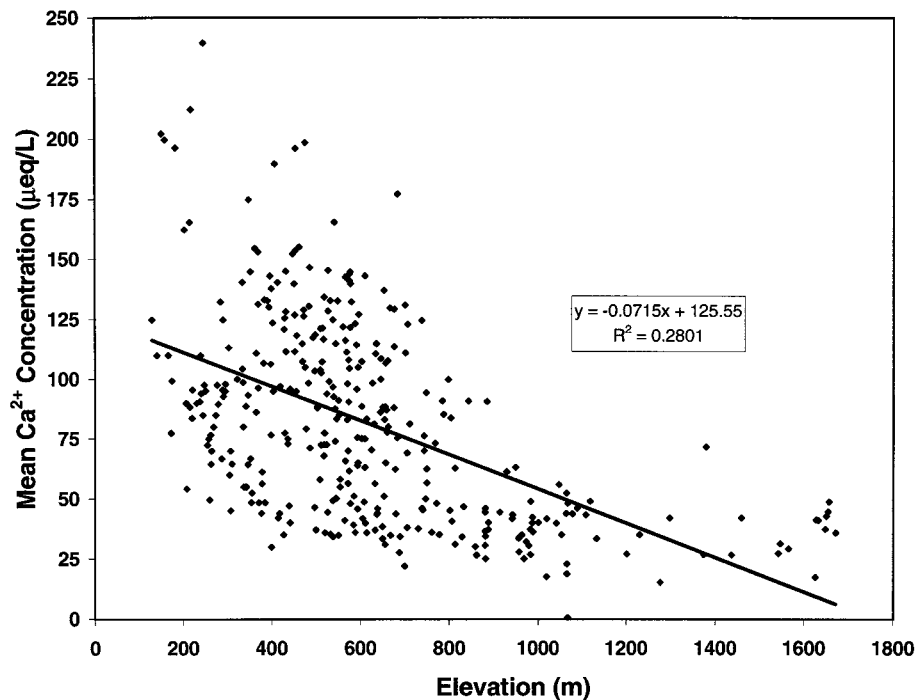


Figure 4.—Mean Ca^{2+} versus elevation.

strong ($R^2 = 0.28$). This is in keeping with the earlier discussion suggesting that the lowest concentrations are associated with small, high-elevation watersheds.

Relationships to Acidic Deposition

Acidic deposition provides hydrogen ions (H^+), which displace cations such as Ca^{2+} on soil surfaces, and sulfate ions (SO_4^{2-}) and nitrate ions (NO_3^-), which keep cations dissolved in soil solution that eventually becomes streamwater. The mobility of Ca^{2+} and its occurrence in streamwater is related to sulfate concentrations (Hornbeck et al. 1997). This relationship holds for the WMNF stations, as shown by a plotting of mean Ca^{2+} versus mean SO_4^{2-} (Fig. 5).

As a result of controls on sulfur emissions imposed in the 1970's, concentrations of SO_4^{2-} in both precipitation and streamwater have been declining over the past several decades. These reductions in SO_4^{2-} have resulted in a gradual decline in base cations such as Ca^{2+} in streamwater (Driscoll et al. 1989; Likens et al. 1996; Lawrence and Huntington 1999). A plotting of monthly means of streamwater Ca^{2+} over time suggests a gradual

reduction in streamwater Ca^{2+} across the WMNF (Fig. 6). Linear regressions fitted to individual stations having at least 2 years of continuous sampling also indicated a downward trend in Ca^{2+} over time for most stations, though several also indicated an opposite trend (Fig. 7).

Other Possibilities

The WMNF geographic information system consists of a variety of layers (e.g., watersheds by stream levels, forest types, and bedrock geology) that could be used to develop relationships and better understand streamwater chemistry. Regional maps and models of atmospheric deposition could be used to determine whether variations in deposition by latitude, longitude, and elevation might affect streamwater chemistry. Also, there are a variety of research results on the impacts of natural and human disturbances on streamwater chemistry on the WMNF (see www.hbrook.sr.unh.edu for a bibliography on Hubbard Brook Ecosystem Studies). These results could be contrasted with the database to predict how background levels for streamwater chemistry would change following disturbances.

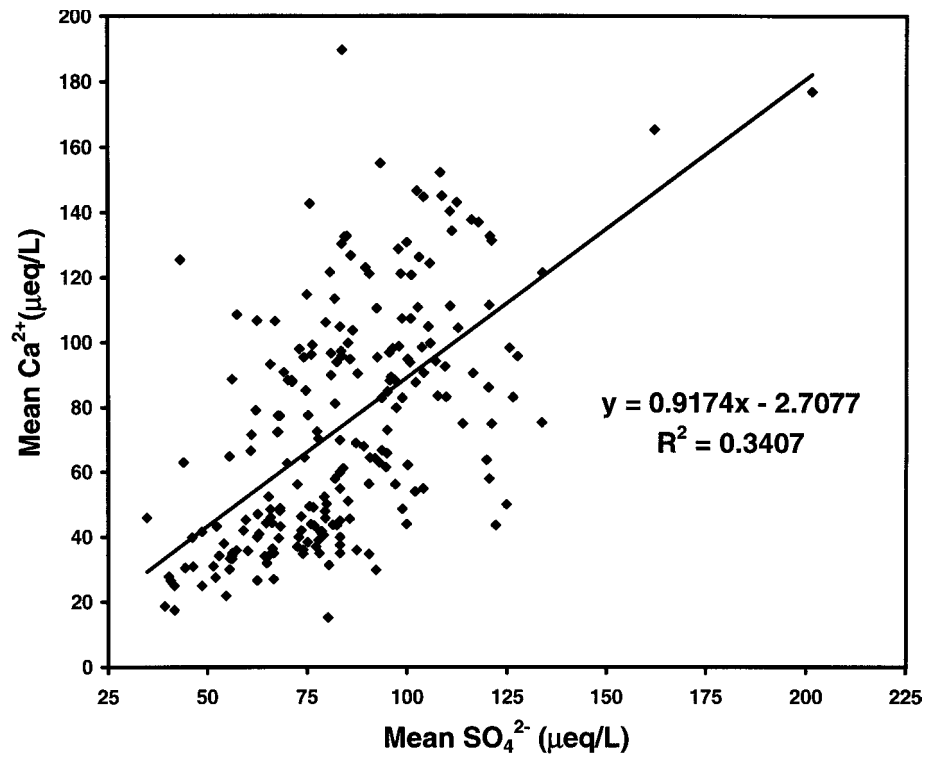


Figure 5.—Relationship of streamwater Ca^{2+} to SO_4^{2-} .

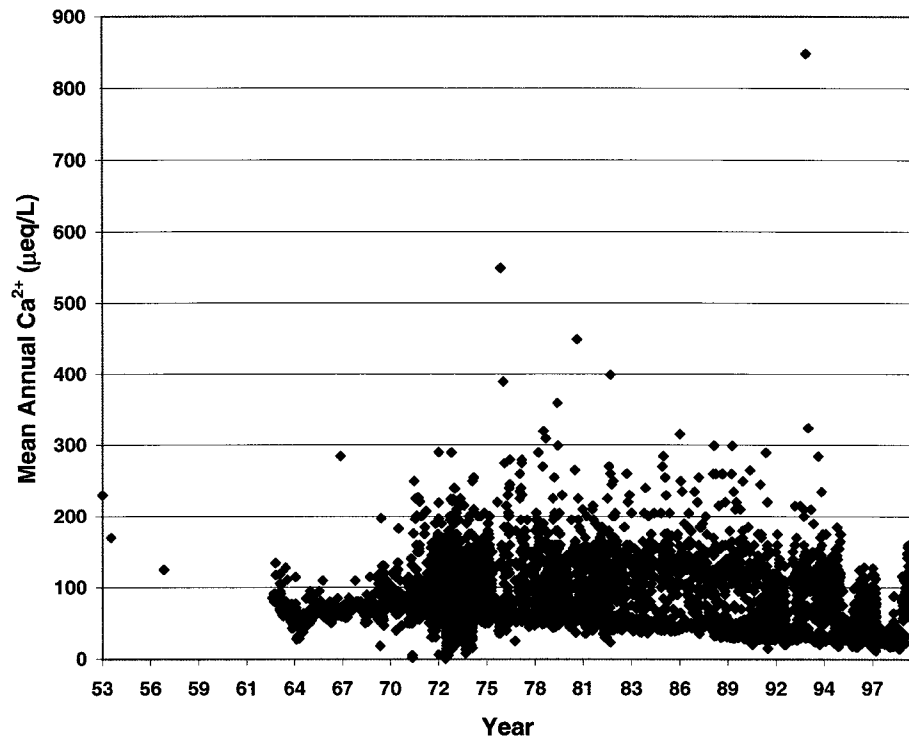


Figure 6.—Mean annual Ca^{2+} for all stations over time.

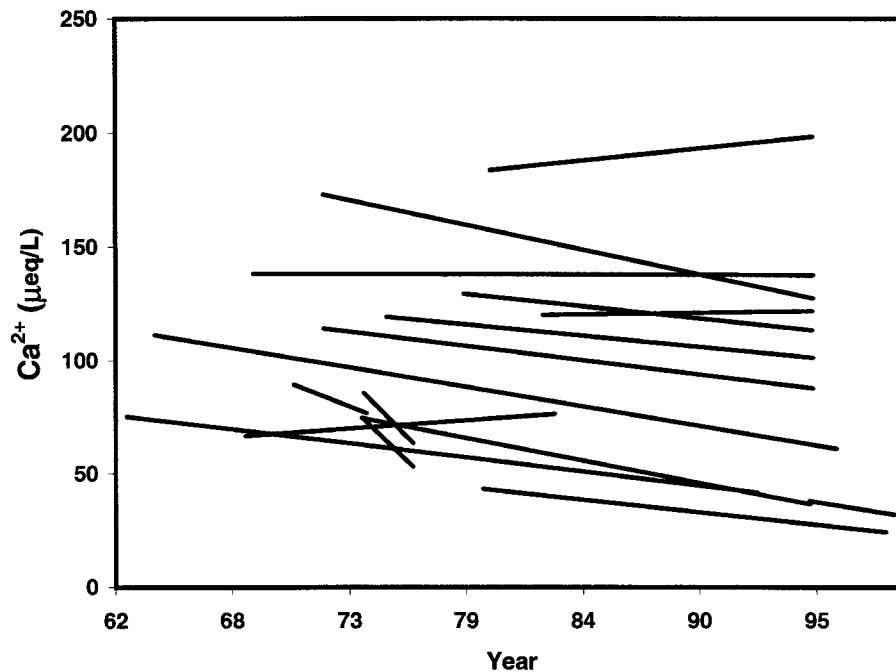


Figure 7.—Regression relationships of Ca^{2+} versus time for selected stations.

Acknowledgement

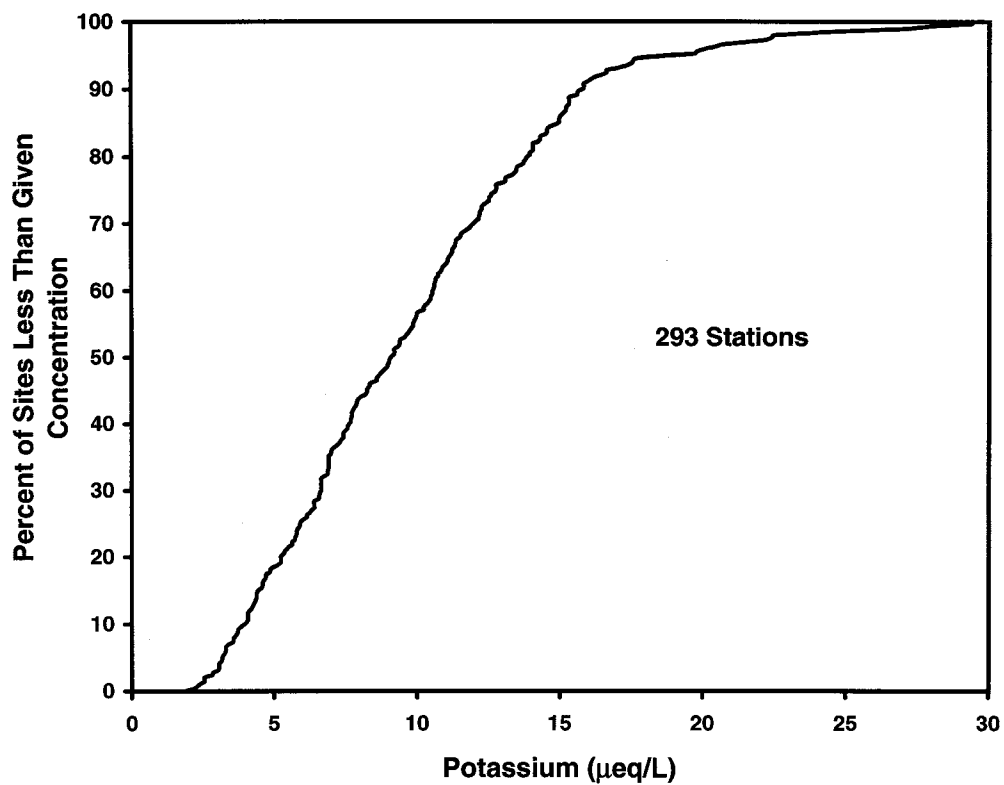
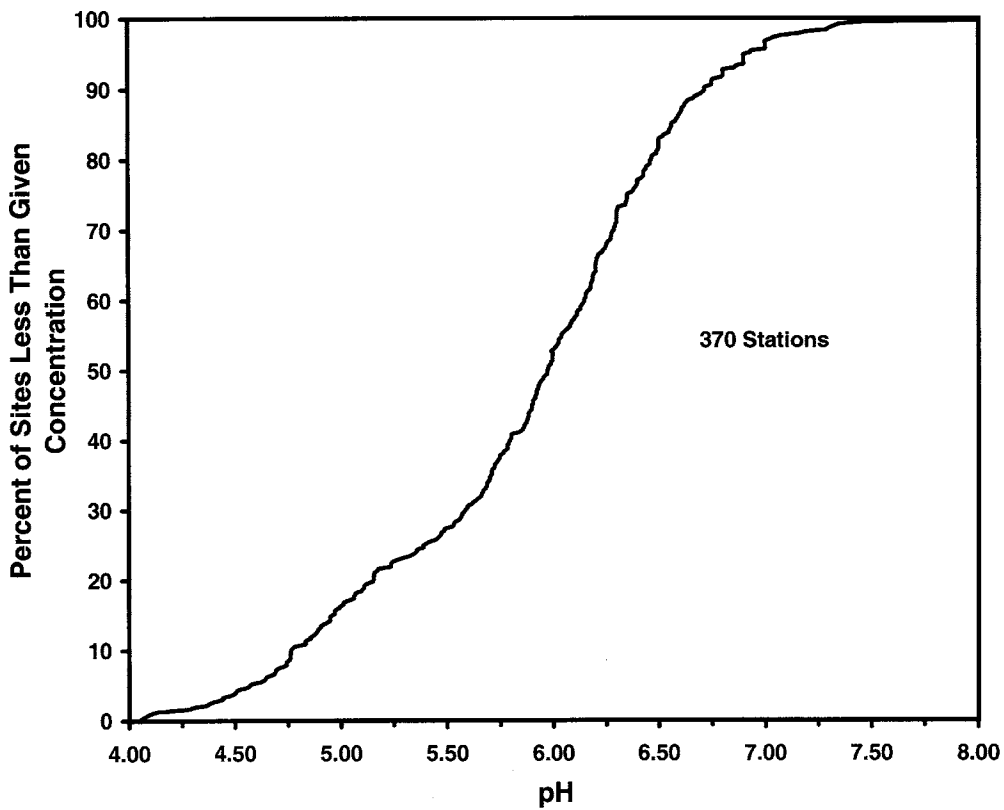
In addition to the many individuals from various organizations who provided data for this report, Steve Jones, Forest Technician on the Ammonoosuc Ranger District, White Mountain National Forest, collected many of the samples included in the database, and assisted in retrieving data from STORET. Ken Dudzik, USDA Forest Service, provided the cover design.

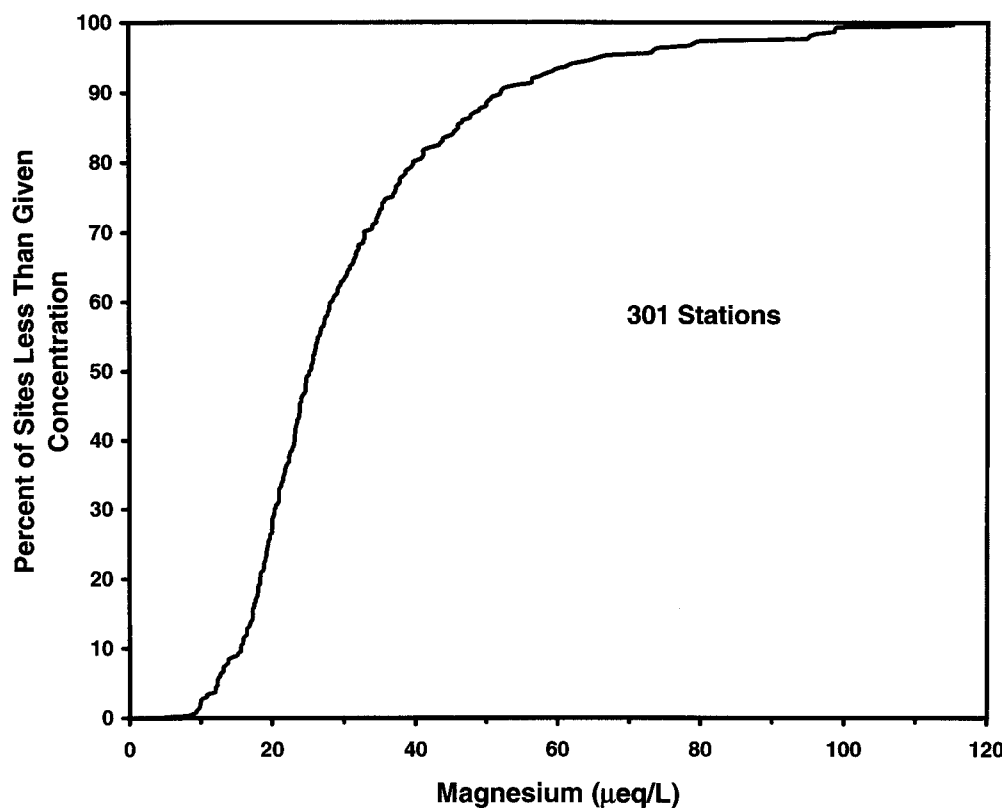
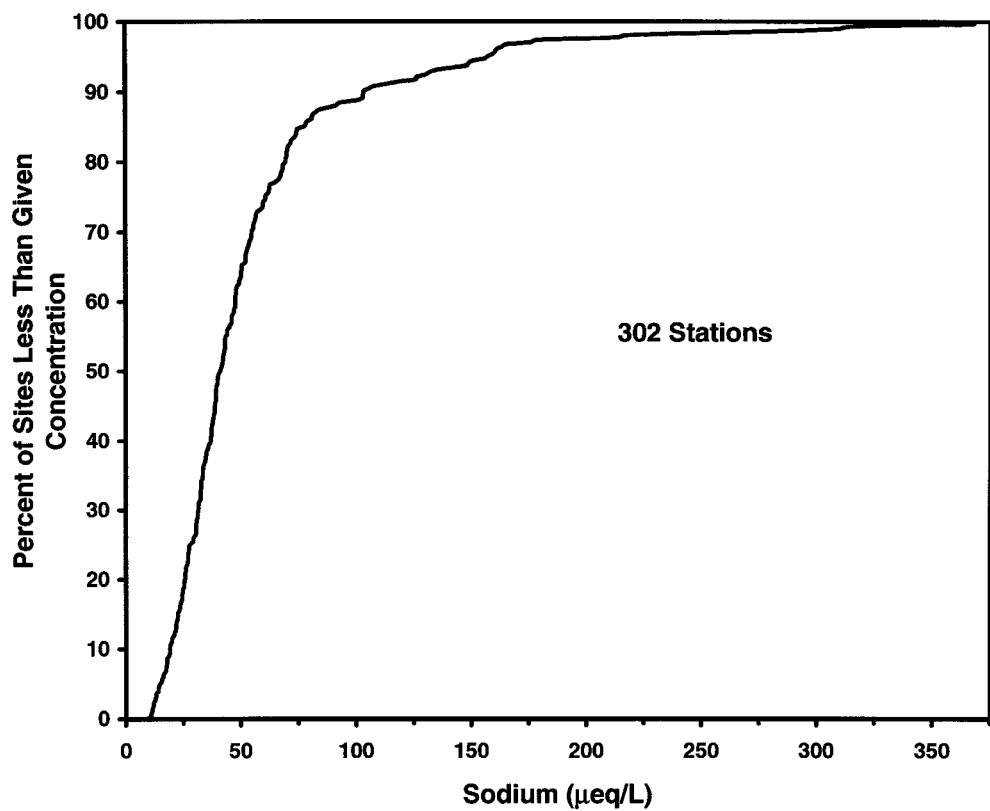
Literature Cited

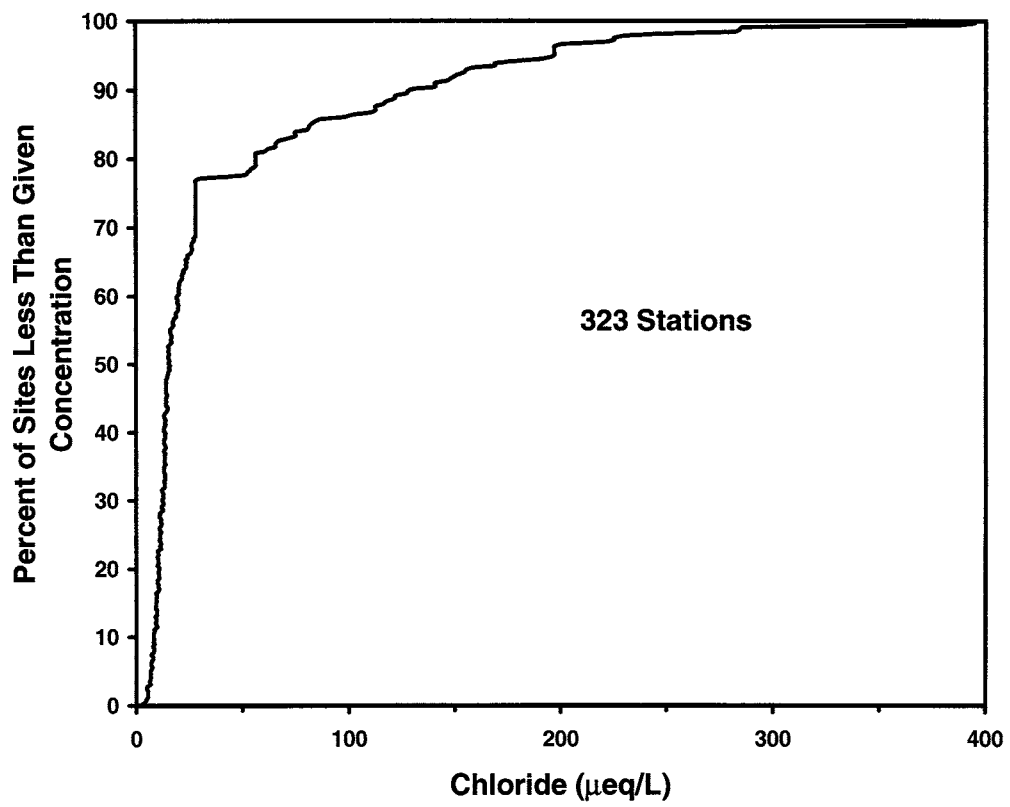
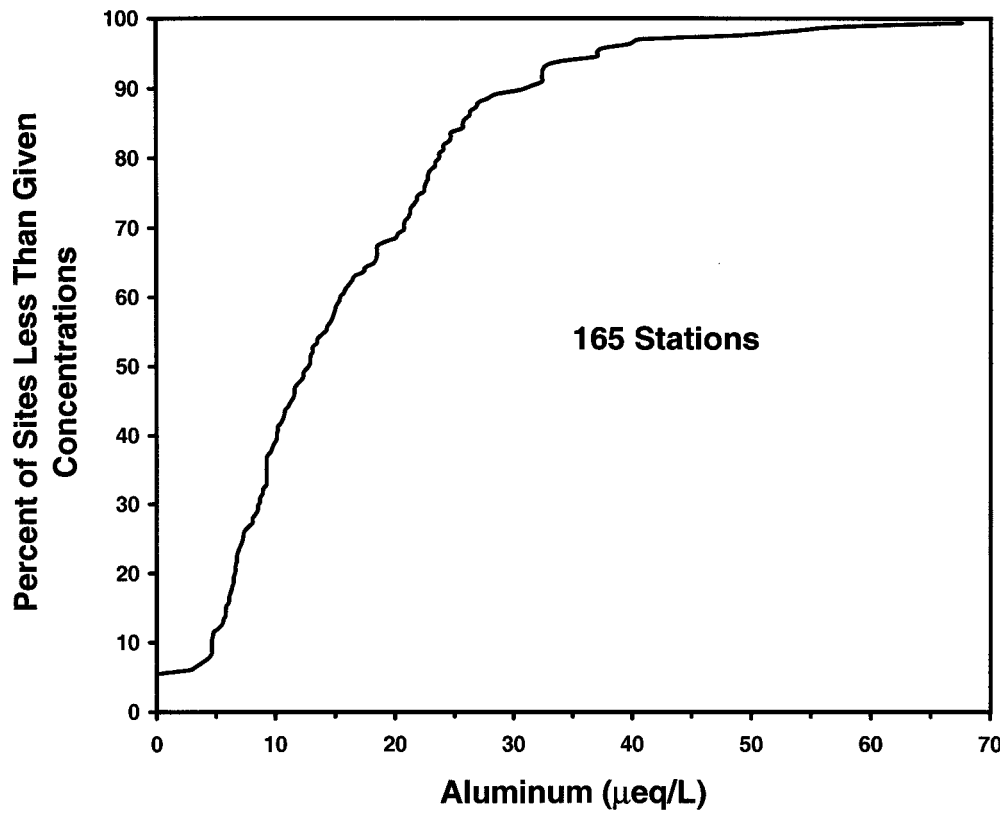
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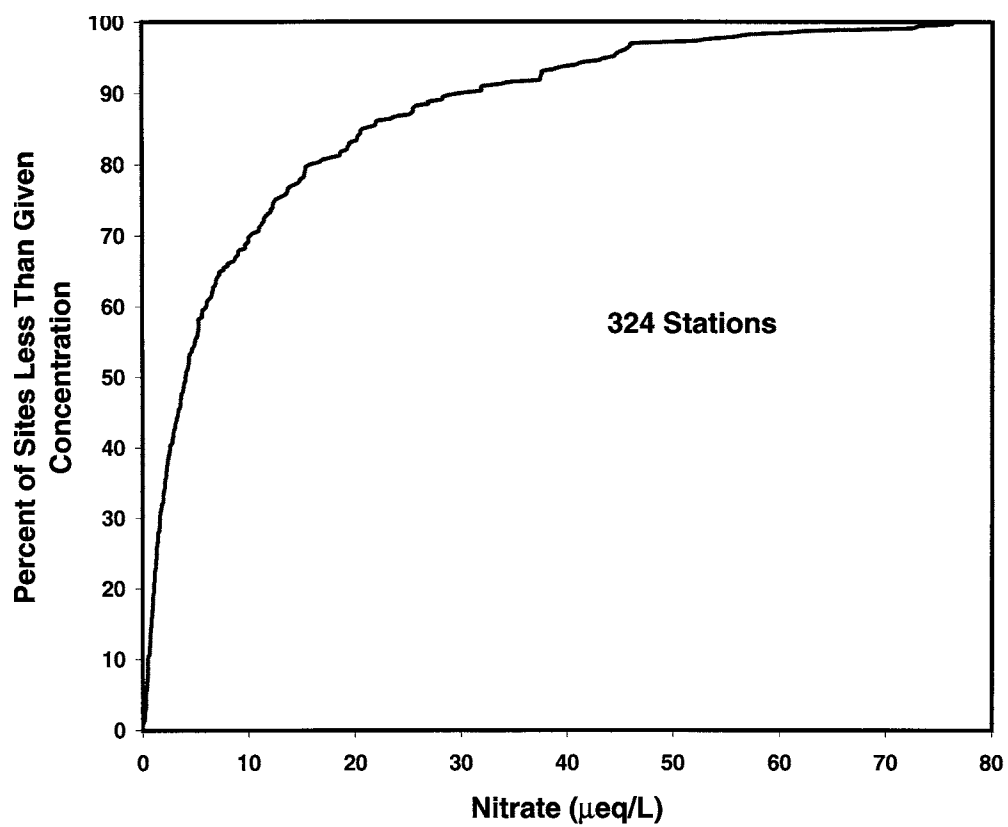
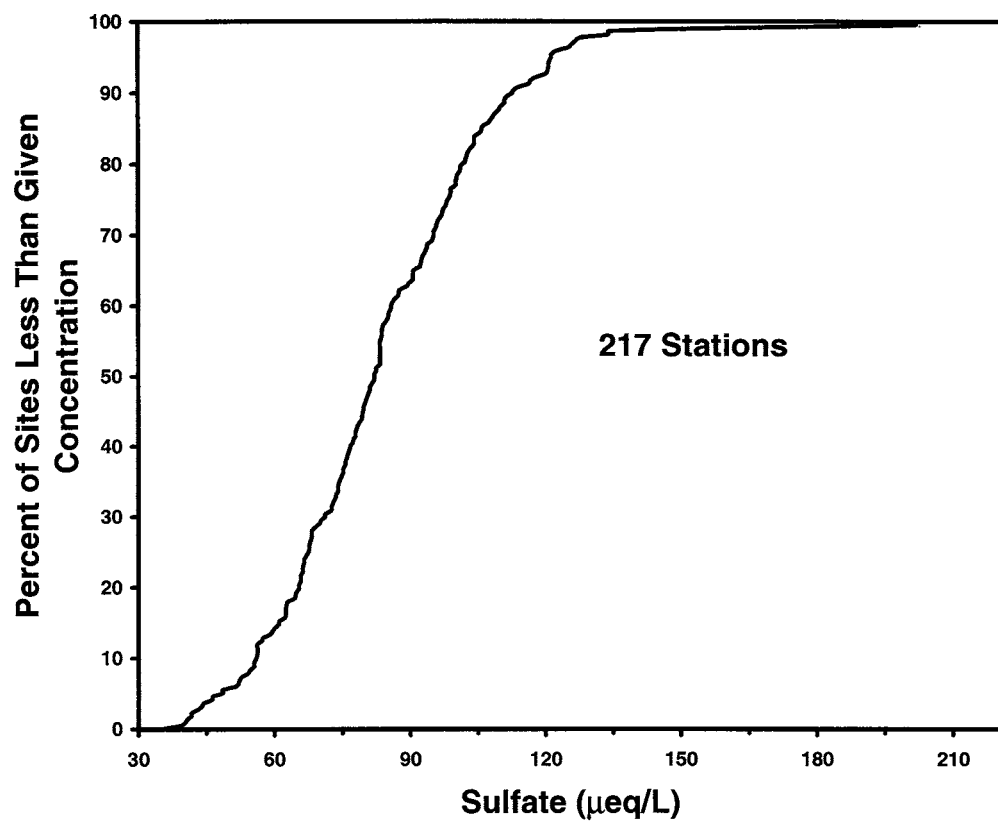
Appendix

Cumulative Frequency Plots for pH, Potassium, Sodium, Magnesium, Aluminum, Chloride, Sulfate, and Nitrate







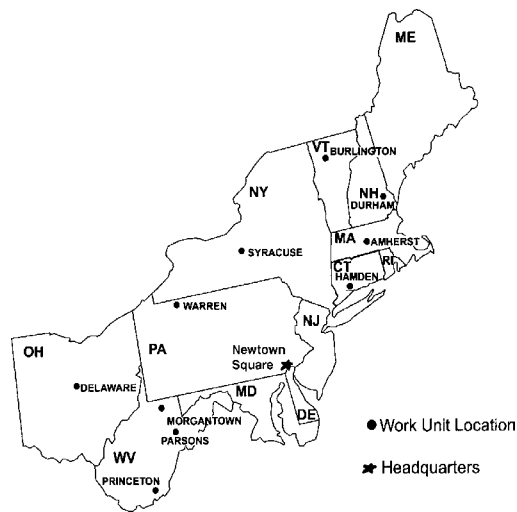


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Keywords: water quality, atmospheric deposition, pH, calcium, sulfate





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